

Unit 04:  
Enzyme-Action of Catalase

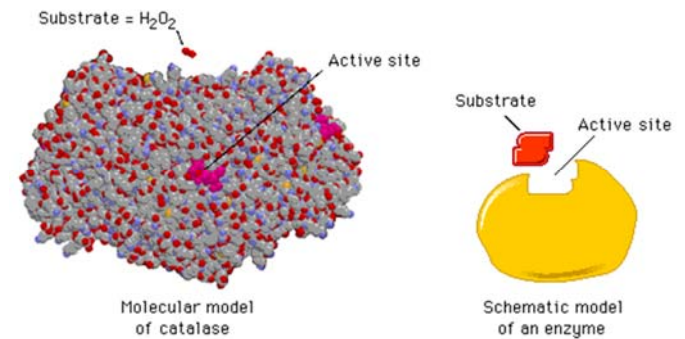
Introduction

REVIEW ON  
ENZYME

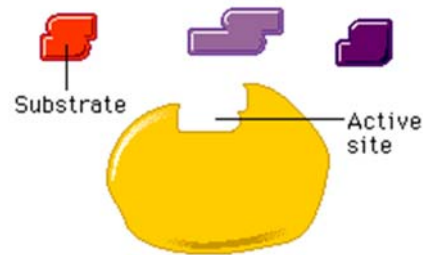
Enzyme

- What is enzyme?
  - Mostly protein**
  - Biological catalyst**
  - Catalyst speeds up chemical reaction **without being used up or altered.****
  - Binds specific substrate only at an active site**

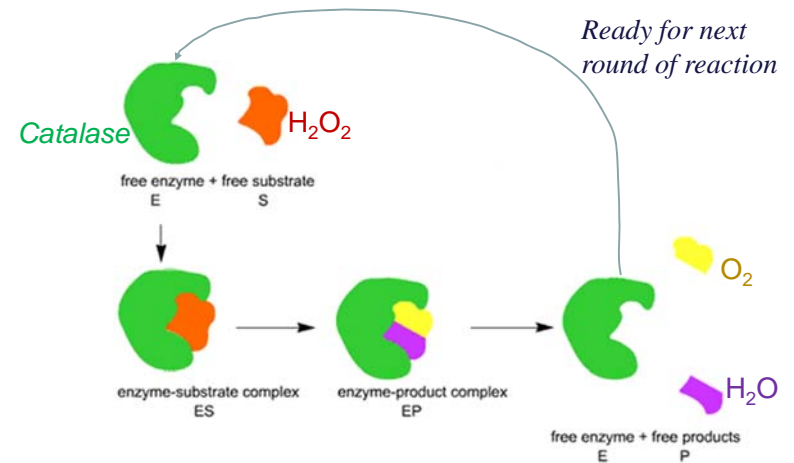
Enzyme: Active site & Substrate



## Specificity of active site & substrate



## Catalytic Reaction of Catalase



## Watch at home

- Animations of Enzyme Action:

[http://highered.mcgraw-hill.com/sites/0072495855/student\\_view0/chapter2/animation\\_how\\_enzymes\\_work.html](http://highered.mcgraw-hill.com/sites/0072495855/student_view0/chapter2/animation_how_enzymes_work.html)



<http://www.youtube.com/watch?v=V4OPO6JQLOE>



## Laboratory Objectives

- Investigate Catalase Activity Based on Scientific Method:

1. **Observation (We know that):**
  - Catalase catalyzes (speeds up) the biochemical reaction to break down hydrogen peroxide into oxygen and water.
  - Catalase works best at an optimum temperature and pH.
2. **Question:**
  - How do the following factors affect the rate of catalase activity (i.e., the speed of reaction)?
    - » Enzyme conc.
    - » Substrate conc.
    - » pH
3. **Hypotheses:**
  - The higher conc. of enzyme or substrate increases the rate of catalase activity.
  - At an optimum pH, the rate of catalase activity is the highest.

## Laboratory Objectives- cont'd

### 4. Predictions:

- If the higher conc. of enzyme or substrate increases the rate of catalase activity, the product ( $O_2$ ) will be formed faster, and thus the filter disc will float faster.
- At an optimum pH, the rate of catalase activity is the highest.
- If the rate of catalase activity is the highest at an optimum pH, the filter disc will float the fastest at the optimum pH.

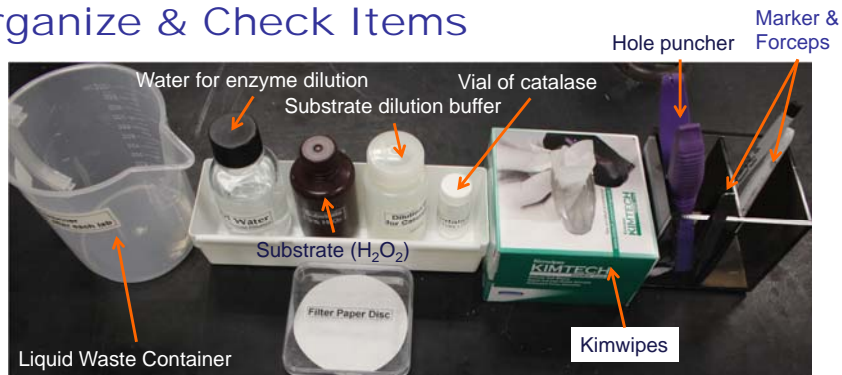
### 5. Conducting Controlled Experiment:

- We will test the hypotheses by measuring time for a filter paper disc to float at different concentrations of catalase and  $H_2O_2$ , and different pH's.

Controlled Experiments

## MEASURING THE RATES OF ENZYME REACTIONS

## Organize & Check Items



- Catalase and  $H_2O_2$  to be provided at the beginning of the lab to equilibrate to room temperature.
- Avoid cross-contamination.
- Keep the marker tip-side down.
- Empty and return the following bottles to lab instructor:
  - ✓ substrate bottle, dilution buffer bottle, enzyme vial.
- Organize remaining items as shown after exercise.

## Front Supply Table

• Left

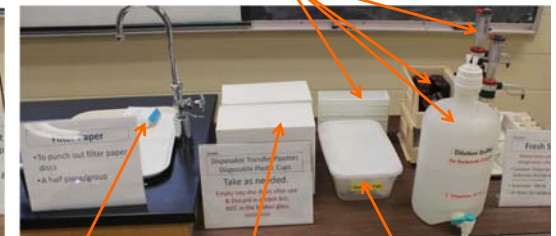
pH 3 pH 5 pH 7



- Buffered  $H_2O_2$  substrates are ready for reaction. (Contains both  $H_2O_2$  & pH buffer.)
- Use 10 ml of each pH

• Right

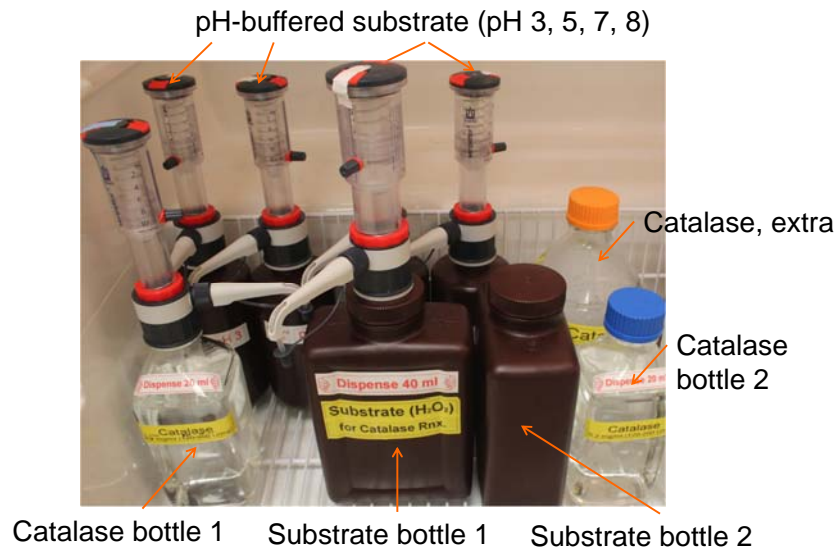
Fresh solution distribution for instructor



Filter Paper for disc preparation Disposable transfer pipettes Disposable Plastic cups

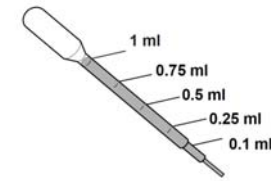
- ✓ Ask lab instructor for refill of dilution buffer, substrate or enzyme.
- ✓ Use fresh cups for each experiment.

## In Refrigerator (for Instructor only)

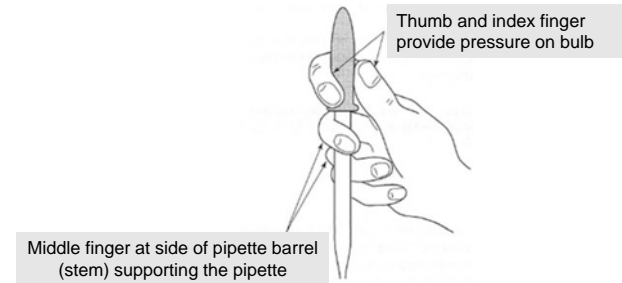


## Using a transfer pipette (Pasteur pipet)

Volume marks on a transfer pipette (Pasteur pipet)



Holding a transfer pipette



## Activity A:

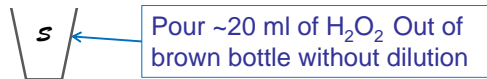
### Enzyme Concentration

Enzyme Conc. (Units/ml)	Dil. Buffer to add (ml)	Catalase to add (ml)
0	4	0
25	3	1
50	2	2
75	1	3
100	0	4

Prepare 5 cups of catalase at different conc.



Prepare 1 cup of Substrate ( $H_2O_2$ )



## Measuring Reaction Time

- Use forceps to grab a disc at an edge. DO NOT hold with fingers. DO NOT deform the disc.
- Dip one filter disc at a time
- Blot excess enzyme solution on a piece of Kimwipes. Be consistent so that each disc contains similar amt. of enzyme
- Measure time at one catalase conc. at a time
- Calculate the rate of reaction using the following formula:

$$\frac{1}{\text{Time required to float}}$$

## Activity B: Substrate Concentration

Prepare 1 cup of Enzyme



Pour ~2-3 ml of **FRESH** undiluted catalase solution (100 U/ml) out of enzyme bottle

H <sub>2</sub> O <sub>2</sub> Conc. (%)	4% H <sub>2</sub> O <sub>2</sub> to add (ml)	Dilution Buffer to add (ml)
0	0	8
1.0	2	6
2.0	4	4
4.0	8	0

Prepare 4 cups of Substrate (H<sub>2</sub>O<sub>2</sub>) at different conc.



## Measuring Reaction Time

(same as in Activity A)

- Use forceps to grab a disc at an edge. DO NOT hold with fingers. DO NOT deform the disc.
- Dip one filter disc at a time
- Blot excess enzyme solution on a piece of Kimwipes. Be consistent so that each disc contains similar amt. of enzyme
- Measure time at one catalase conc. at a time
- Calculate the rate of reaction using the following formula:

$$\frac{1}{\text{Time required to float}}$$

## Activity C: Effect of pH

1 cup of Enzyme



Pour ~2-3 ml of **FRESH** undiluted enzyme solution (100 U/ml) out of blue capped tube

- Obtain ~10 ml of each buffered substrate in separate cups from brown square bottles on the front supply table.
- The buffered H<sub>2</sub>O<sub>2</sub> solutions are ready for reaction. DO NOT add any solution; just place catalase-soaked filter paper discs.

Obtain 4 cups of buffered Substrate (H<sub>2</sub>O<sub>2</sub>) of different pH



## Measuring Reaction Time

(same as in Activity A & B)

- Use forceps to grab a disc at an edge. DO NOT hold with fingers. DO NOT deform the disc.
- Dip one filter disc at a time
- Blot excess enzyme solution on a piece of Kimwipes. Be consistent so that each disc contains similar amt. of enzyme
- Measure time at one catalase conc. at a time
- Calculate the rate of reaction using the following formula:

$$\frac{1}{\text{Time required to float}}$$

## Completion of the lab

- Empty & dispose of used plastic cups in the trash can.
- Empty & **return** the catalase tube, bottles for 4%  $H_2O_2$  and dilution buffer so that your lab instructor can refill them for the next lab session.  
Not required to wash unless they are contaminated with enzyme.
- **Empty, rinse & return** the Liquid Waste Container on your lab table.

## Organize & Check Items



- Catalase and  $H_2O_2$  to be provided at the beginning of the lab to equilibrate to room temperature.
- Avoid cross-contamination.
- Keep the marker tip-side down.
- Empty and return the following bottles to lab instructor:
  - ✓ substrate bottle, dilution buffer bottle, enzyme vial.
- Organize remaining items as shown after exercise.